New Studies on [2+3] Cycloadditions of Thermally Generated N-Isopropyland N-(4-Methoxyphenyl)-Substituted Azomethine Ylides

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The thermal reaction of 1-substituted 2,3-diphenylaziridines **2** with thiobenzophenone (**6a**) and 9*H*-fluorene-9-thione (**6b**) led to the corresponding 1,3-thiazolidines (*Scheme 2*). Whereas the *cis*-disubstituted aziridines and **6a** yielded only *trans*-2,4,5,5-tetraphenyl-1,3-thiazolidines of type **7**, the analogous reaction with **6b** gave a mixture of *trans*- and *cis*-2,4-diphenyl-1,3-thiazolidines **7** and **8**. During chromatography on SiO₂, the *trans*-configured spiro[9*H*-fluorene-9,5'-[1,3]thiazolidines] **7c** and **7d** isomerized to the *cis*-isomers. The substituent at N(1) of the aziridine influences the reaction rate significantly, *i.e.*, the more sterically demanding the substituent the slower the reaction. The reaction of *cis*-2,3-diphenylaziridines **2** with dimethyl azodicarboxylate (**9**) and dimethyl acetylenedicarboxylate (**11**) gave the *trans*-cycloadducts **10** and **12**, respectively (*Schemes 3* and 4). In the latter case, a partial dehydrogenation led to the corresponding pyrroles. Two stereoisomeric cycloadducts, **15** and **16**, with a *trans*-relationship of the Ph groups were obtained from the reaction with dimethyl fumarate (**14**; *Scheme 5*); with dimethyl maleate (**17**), the expected cycloadducts **18** together with the 2,3-dihydropyrrole **19** was obtained (*Scheme 6*). The structures of the cycloadducts **7b**, **8a**, **15b**, and **16b** were established by X-ray crystallography.

Introduction. – Recently, we focused our attention on 1,3-dipolar cycloadditions of thiocarbonyl dipolarophiles with azomethine ylides by different procedures for their generation (cf. [1] and refs. cit. therein). One of the most frequently used procedures is the thermal, stereoselective ring opening of N-substituted aziridines [2][3]. This method was explored to synthesize 1,3-thiazole derivatives in a regio- and stereoselective manner [4–6]. In some cases, drastic reaction conditions were necessary, which led to the formation of unexpected products. For example, the reaction of 2,2,4,4-tetramethyl-3-thioxocyclobutanone (1) with dimethyl 1,3-diphenylaziridine-2,2-dicarboxylate (2a) gave, among other products, dimethyl 1,3-thiazolidine-2,2-dicarboxylate (4, which is formally a cycloadduct of the azomethine ylide with dimethyl-thioketene (Scheme 1) [7]. Furthermore, reactions of 1 with 1-methyl- and 1-isopropyl-2,3-diphenylaziridine 2b and 2c, respectively, led to the expected 1,3-thiazolidines 5a and 5b, respectively, but the reaction with 2c was considerably slower and accompanied by significant decomposition (Scheme 1). This different reactivity can be explained by

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²⁾ Part of the diploma thesis of R. S., University of Łódź.

Scheme 1

Ph
$$CO_2Me$$
 CO_2Me
 C

the assumption of a severe steric interaction between the i-Pr substituent and a Ph group in the intermediate azomethine ylide.

To gain more insight into the influence of the *N*-substituent of 2,3-diphenylaziridines on the reactivity of the azomethine ylides generated therefrom, we compared the reactions of **2b** and **2c** as well as those of *trans*-1-benzyl-2,3-diphenylaziridine (**2d**) and *cis*-1-(4-methoxyphenyl)-2,3-diphenylaziridine (**2e**). For these studies, aromatic thioketones, dimethyl acetylenedicarboxylates, dimethyl fumarate and maleate, as well as dimethyl azodicarboxylate were selected as dipolarophiles.

Results and Discussion. – The thermal reaction of **2b** with thiobenzophenone (**6a**) was reported to occur smoothly (4 h) and gave the expected 1,3-thiazolidine **7a** in high yield [8]. Under analogous conditions, **2c** and **6a** yielded, after 24 h, **7b** as the sole product (49%). The predicted *trans*-relationship of the Ph substituents at C(2) and C(4) was confirmed by X-ray crystallography (*Fig. I*). Under the same conditions, **2d** and **6a** gave, after 7-h heating, the *cis*-configured cycloadduct **8c** in 44% yield.

To compare the reactivities of the aziridines with a Me and i-Pr group, respectively, at the N-atom, a mixture of equimolar amounts of 2b, 2c, and 6a in toluene was heated under reflux for $3 \, h^3$). The ratio 7a/7b was established by 1H -NMR spectroscopy: the singlets from H-C(2) and H-C(4) of the two adducts (7a: 5.40 and 5.13 ppm; 7b: 5.78 and 5.68 ppm) showed intensities with a ratio of ca. 3:1. Based on kinetic evidence, Huisgen et al. concluded that, in the case of N-benzyl aziridine 2d and dimethyl fumarate, the rate of the formation of the cycloadduct depends exclusively on the rate of the ring opening [10a]. In our system, the interception of the azomethine ylides by the 'superdipolarophile' 6a is assumed to occur very fast, and, therefore, the higher yield of 7a indicates that 2b undergoes the electrocyclic ring opening faster than 2c. The bulkier i-Pr substituent apparently reduces the rate of the conrotatory ring opening.

³⁾ After this time, the blue color of **6a** had disappeared completely.

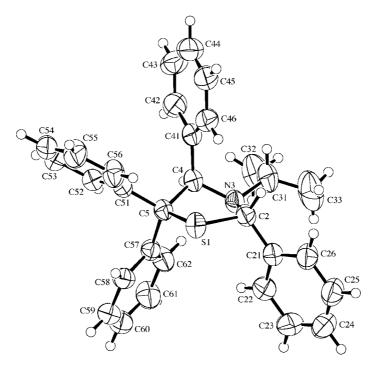


Fig. 1. ORTEP Plot [9] of the molecular structure of one of the two symmetry-independent molecules of **7b** (40% probability ellipsoids, arbitrary numbering of atoms)

Analogously, the reaction of **2b** with 9*H*-fluorene-9-thione (**6b**) gave **7c** and traces of 8a (1H-NMR; Scheme 2). After chromatography (SiO₂), only 8a was obtained in 35% yield, i.e., an isomerization $7c \rightarrow 8a$ took place during chromatographic workup. The cis-configuration of 8a has been unambigously confirmed by X-ray crystallography (Fig. 2). In the case of **2c** and **6b**, again a mixture of two cycloadducts was formed. The product isolated in 42% yield after chromatographic separation⁴) corresponded to the minor component of the reaction mixture. Based on their spectroscopic data, both products are 3-isopropyl-1,3-thiazolidines, which differ in the relative configuration of the Ph ring at C(2) and C(4). By analogy to the reaction with **2b**, we assign the *trans*configuration to the major product 7d, and the cis-configuration to 8b. Thus, the primarily formed trans-configured products 7c and 7d, i.e., the spirofluorenyl compounds, undergo a smooth isomerization to give the cis-isomers 8a and 8b, respectively, during chromatographic workup⁵). Attempted separations by fractional crystallization always led to mixtures of the isomers. When a sample of recrystallized material containing 7d and 8b (9:1 mixture) was dissolved in CDCl₃ in the presence of catalytic amounts of CF₃COOH, the isomerization to 8b was complete after 3 d at

⁴⁾ The ratio 7d/8b of the isolated material depended on how long the mixture was exposed to SiO₂. After 12 h, the isomerization was complete.

⁵⁾ It is worth mentioning that an analogous isomerization was not observed in the case of 7b.

Scheme 2

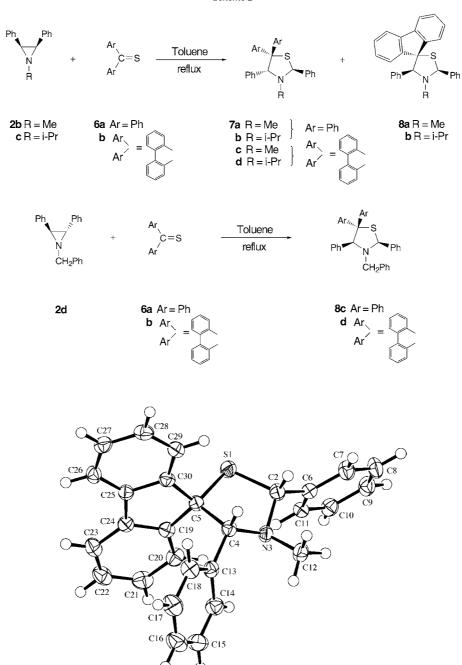


Fig. 2. ORTEP Plot [9] of the molecular structure of one of the two symmetry-independent molecules of **8a** (50% probability ellipsoids, arbitrary numbering of atoms)

room temperature⁶). This isomerization can be rationalized by a ring-opening/ring-closure process in which a thiocarbonylium ion (*via* cleavage of the C(2)-N(3) bond) or an azomethinium ion (*via* cleavage of the S(1)-C(2) bond) is the intermediate.

The isomerizations of **7c** and **7d** to **8a** and **8b**, respectively, indicate that the *cis*-configured 2,4-diphenyl-1,3-thiazolidines are the thermodynamically more stable stereoisomers. This observation was confirmed in an additional experiment in which *trans*-aziridine **2d** was reacted with **6b**. In this case, only one cycloadduct, **8d** (*Scheme* 2), which was stable during chromatography (SiO₂), was found in the mixture. The similarity of its NMR spectra with those of **8a** and **8b** indicated the expected *cis*-configuration.

When a 1:1 mixture of 2c and dimethyl azodicarboxylate (9) was heated in toluene for $15 \, h^7$), the cycloadduct 10a (*Scheme 3*) was obtained, which was purified by chromatography on $Al_2O_3{}^8$). As an additional precursor of an azomethine ylide, the 1-aryl-substituted aziridine 2c was reacted with 9 in boiling toluene to give 10b (*cf.* [12]). In this case, the reaction was complete after only $2 \, h^9$).

Dimethyl acetylenedicarboxylate (11) is frequently used as an electron-deficient dipolarophile (cf. [13]). The reaction with 2c in refluxing toluene was complete after 8 h. By ¹H-NMR analysis, the products 12a and 13a were detected in a ratio of 85:15 (Scheme 4). The main differences were the presence of a singlet at 5.61 ppm (2 H) and two doublets for the i-Pr group in the case of the major product 12a, whereas 13a showed only one doublet for the i-Pr substituent. Crystallization from MeOH gave 13a, and subsequent chromatographic separation of the mother liquor led to an additional amount of 13a and to 12a. In a second experiment, the mixture was heated for 24 h to yield 13a exclusively.

The higher reactivity of **2e** was confirmed in the reaction with **11**: here, **2e** was completely consumed after just 1 h. Again, two products were obtained, which were identified as **12b** and **13b**. The smooth conversion of 2,5-dihydro-1*H*-pyrroles into 1*H*-

⁶⁾ A similar isomerization of N-methyl- and N-benzyl-2,4,5-triaryl-1,3-oxazolidines was reported to occur thermally or during chromatography on SiO₂ [11b,c].

No 2c could be detected after this time.

First attempts with SiO₂ as the adsorbent led to decomposition of **10a**.

⁹⁾ In 1966, Heine et al. reported the reaction of 1,2,3-triphenylaziridine with 9 in refluxing toluene (4 h) [3b]. No reaction mechanism was proposed, as, at that time, the electrocyclic ring opening of aziridines to azomethine ylides was not known. Neither the configuration of the starting material nor of the product were given.

b $R = 4\text{-MeOC}_6H_4$

pyrroles by spontaneous dehydrogenation in the presence of air-O₂ (autoxidation) or after addition of an oxidizing agent is well-established [14][15].

b $R = 4\text{-MeOC}_6H_4$

Thermal as well as photochemical reactions of 2,3-diarylaziridines with 11 have been reported. Under conditions similar to those described above, trans-1-methyl-2,3diphenylaziridine gave dimethyl cis-2,5-dihydro-1-methyl-2,5-diphenyl-1H-pyrrole-3,4dicarboxylate as the sole product [16]. On the other hand, UV irradiation of trans-1,2,3triphenylaziridine in the presence of 11 led to a mixture of the trans-disubstituted 2,5dihydro-1*H*-pyrrole **12** (R = Ph) and the corresponding 1*H*-pyrrole derivative **13** (R = Ph) Ph) [15]. In some cases, the formation of 2,3-dihydro-1*H*-pyrroles has been observed. For example, thermolysis of 1-methyl-2-(4-methylphenyl)-3-(2-pyridyl)aziridine and 11 was reported to yield a compound of this type [17]. Furthermore, photolysis of trans-1-butyl-2,3-diphenylaziridine and **11** afforded a mixture of all three types of 1*H*-pyrrole derivatives [18]. Apparently, the formation of 2,3-dihydro-1*H*-pyrroles is the result of an isomerization of the initially formed 2,5-dihydro-1H-pyrrole¹⁰).

Both aziridines, 2c and 2e, were reacted with dimethyl fumarate (=dimethyl (E)but-2-enedioate; 14) to yield mixtures of the two diastereoisomeric pyrrolidine-3,4dicarboxylates 15 and 16 (Scheme 5). Whereas the ratio 15a/16a was almost 1:1, it was established as 4:1 in the case of 15b/16b. On the basis of the ¹H-NMR spectra, we proposed that compounds 15 are the (r-2, t-3, c-4, t-5) isomers: H-C(3), H-C(4)absorb as dd at ca. 4.1 and 3.5 ppm in the cases of 16 and 15, respectively. The high-field shift in **15a** and **15b** is a result of the shielding effect of the Ph group at C(2) and C(5). The pure isomers of type 15 and 16 were obtained after chromatographic separation, and, in the case of 15b and 16b, the predicted structures were established by X-ray

$$Scheme 5$$

$$Ph \qquad Ph \qquad MeO_2C \qquad Toluene \qquad MeO_2C \qquad CO_2Me \qquad MeO_2C \qquad CO_2Me \qquad Ph^{1/2}Ph \qquad Ph \qquad Ph^{1/2}Ph \qquad$$

In analogous reactions with 2-benzoyl- and 2-(methoxycarbonyl)-substituted aziridines, the formation of 2,5-dihydro- and/or 2,3-dihydro-1*H*-pyrroles as well as 1*H*-pyrroles was observed [14][19-22].

crystallography (Fig. 3). These results are similar to the ratios reported earlier for the diastereoisomeric pyrrolidinedicarboxylates obtained from the reactions of the corresponding N-benzyl- and N-Ph-substituted aziridines, respectively, and 14 [10] [23].

Dimethyl maleate (17) is known to be less reactive than the above dipolarophiles [13]. In a previous study, however, it has been shown that the respective pyrrolidine was formed *via* interception of the azomethine ylide generated either by thermal ring opening of 2b or by addition of methyl(phenyl)carbene to *N*-benzylidenemethylamine [24]. Unexpectedly, thermolysis of 2c in the presence of 17 gave none of the expected interception products. After heating for 100 h, analysis of the crude mixture showed the presence of small amounts of 15a and 16a, *i.e.*, the products of the reaction with 14 (*Scheme 5*). This result shows that 17 is not reactive enough to trap the intermediate azomethine ylide. Instead, slow isomerization of 17 to 14 takes place.

On the other hand, the reaction of **2e** and **17** in boiling toluene for 24 h led to a mixture of the expected [2+3] cycloadduct **18** and the 2,3-dihydro-1*H*-pyrrole **19**, which is a partially dehydrogenated product of **18**, in a ratio of 5:2 (*Scheme* $6)^{11}$). The partial dehydrogenation of **18** could, in principle, lead to three different products, one of them being the already described **12b**. The elucidation of the structure of **19** was based on the disappearence of the high-field-shifted dd for H-C(4) (δ 3.40 ppm) and the signal of one of the H-atoms ascribed to H-C(2) and H-C(5) (δ 5.60/5.84 ppm) in **18**.

In summary, the present study shows that the N-substituent of cis-2,3-diphenylaziridines 2 significantly influences their reactivity as precursors of the corresponding azomethine ylides. In all cases, the N-isopropyl derivative showed the lowest reactivity, but, in spite of this, [2+3] cycloaddition products with C=S, N=N, and electron deficient C=C dipolarophiles were obtained. The formation of a cycloadduct was also observed in the case of dimethyl fumarate, whereas no reaction occurred with dimethyl maleate. The influence of the i-Pr substituent compared with that of a Me or 4-MeOC₆H₄ group is remarkable. It seems likely that steric as well as electronic effects are responsible for the reduced stability of the ylide and for its slower formation. Furthermore, steric hindrance may be the reason for the more sluggish cycloaddition reaction that generates five-membered heterocycles.

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¹¹⁾ In an earlier experiment with cis-1,2,3-triphenylaziridine and 17, Huisgen et al. described the cycloadduct analogous to 18, but no secondary product of type 19 was mentioned [10b].

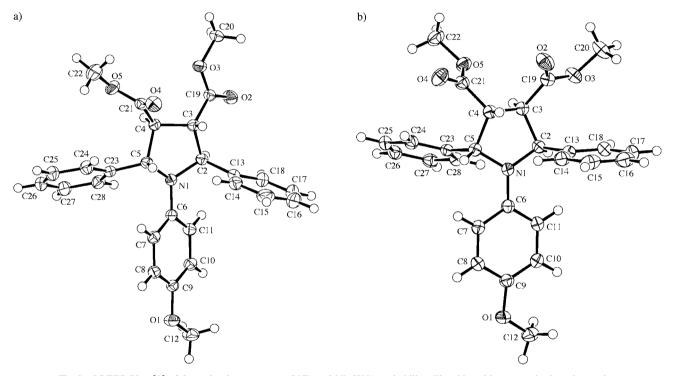


Fig. 3. ORTEP Plots [9] of the molecular structures of 15b and 16b (50% probability ellipsoids; arbitrary numbering of atoms)

Experimental Part

- 1. General. See [25].
- 2. Starting Materials. cis-1-Methyl-2,3-diphenylaziridine (**2b**) and cis-1-isopropyl-2,3-diphenylaziridine (**2c**) were prepared by Gabriel's method according to [11], trans-1-benzyl-2,3-diphenylaziridine (**2d**) and cis-1-(4-methoxyphenyl)-2,3-diphenylaziridine (**2e**) were synthesized according to procedures described by Huisgen et al. [10] [23]. Thiobenzophenone (**6a**) was obtained from benzophenone by treatment with Lawesson's reagent [26], and 9H-fluorene-9-thione (**6b**) was prepared according to [27] by thionation of 9H-fluorenone with a H₂S/HCl mixture at 0-5° in EtOH. Other reagents used in the present study were commercially available. Reported yields refer to isolated products.
- 3. Thermal Reactions of Aziridines **2b 2d** with Dipolarophiles. 3.1. Reaction of **2c** with **6a**. A soln. of **2c** (261 mg, 1.1 mmol) and freshly purified, blue **6a** (198 mg, 1.0 mmol) dissolved in toluene (4 ml) was heated under reflux for 24 h. After this time, the blue color of **6a** disappeared; the mixture was cooled to r.t., and the solvent was removed *in vacuo*. The thick, oily residue was analyzed by ¹H-NMR, and only one product with two *s* for CH at 5.68 and 5.78 ppm was detected. Isolation of the product was achieved chromatographically (SiO₂; petroleum ether with increasing amounts of CH₂Cl₂). The ¹H-NMR spectrum of the isolated material corresponded to that recorded for the crude mixture.

trans-3-Isopropyl-2,4,5,5-tetraphenyl-1,3-thiazolidine (**7b**). Yield: 213 mg (49%). Colorless crystals. M.p. $99-101^{\circ}$ (dec.; blue color; MeOH). IR (KBr): 1601m, 1492s, 1454s, 1362m, 1239s, 1181s, 749s, 729s, 696vs. 1 H-NMR: 0.58, 1.07 (2d, J = 6.6, Me_2 CH); 2.99 (m, Me_2 CH); 5.68, 5.78 (2s, H – C(4), H – C(2)); 6.84 – 7.67 (m, 20 arom. H). 13 C-NMR: 19.3, 21.7 (2q, Me_2 CH); 48.8 (d, Me_2 CH); 68.9 (s, C(5)); 69.9, 73.9 (2d, C(4), C(2)); 125.7, 125.9, 126.8, 127.2, 127.4, 127.5, 127.7, 128.0, 128.1, 129.0, 130.1 (11d, 20 arom. CH); 140.3, 142.5, 143.4, 150.7 (4s, 4 arom. C). CI-MS (NH₃): 436 (100, [M + 1] $^+$), 230 (7), 208 (7). Anal. calc. for C₃₀H₂₉NS (435.63): C 82.72, H 6.71, N 3.22, S 7.36; found: C 82.74, H 6.65, N 3.26, S 7.46.

3.2. Reaction of 2d with 6a. A soln. of 2d (314 mg, 1.1 mmol) and 6a (198 mg, 1.0 mmol) in toluene was heated under reflux for 7 h. The analysis of the crude mixture and isolation of the product were carried out as described in 3.1.

cis-3-Benzyl-2,4,5,5-tetraphenyl-1,3-thiazolidine (**8c**). Yield: 213 mg (44%). Colorless crystals. M.p. 121–123° (MeOH). IR (KBr): 1500s, 1460s, 1450s, 1220m, 1150m (br.), 1090m, 1050m, 780s, 770s, 710vs. 1 H-NMR: 3.70, 3.76 (*AB*, *J* = 14.0, CH₂); 5.17, 5.25 (2s, H–C(4), H–C(2)); 6.92–7.74 (m, 20 arom. H). 13 C-NMR: 51.8 (t, CH₂); 68.4 (s, C(5)); 69.5, 72.5 (2d, C(4), C(2)); 126.2, 126.3, 127.0, 127.2, 127.8, 127.9, 128.1, 128.3, 128.7, 130.0, 130.2, 130.7 (12d, 20 arom. CH); 134.9, 138.5, 139.9, 142.7, 147.1 (5s, 5 arom. C). EI-MS: 285 (76), 195 (21), 194 (100), 178 (6), 121 (8), 91 (20). Anal. calc. for C₃₄H₂₉NS (483.68): C 84.43, H 6.04, N 2.89, S 6.63; found: C 84.59, H 6.04, N 2.90, S 6.39.

3.3. Reactions of **2b**, **2c**, and **2d** with **6b**. A soln. of 1.1 mmol of the corresponding aziridine **2** in 2 ml of toluene was heated under reflux. To these mixtures, a soln. of **6b** (196 mg, 1 mmol) in toluene (2 ml) was added in three portions in intervals of *ca*. 20 min.

Reaction with 2b. The reaction was complete after 2 h. After careful evaporation of the solvent, the thick oily residue was analyzed by ¹H-NMR (presence of two s for CH at 6.06 and 4.98 ppm, resp.). After chromatographic workup (SiO₂; petroleum ether with increasing amounts of CH₂Cl₂), a colorless, thick oil was isolated, which subsequently solidified in the refrigerator. Crystallization from MeOH afforded an anal. pure sample of 8a.

trans-3'-Methyl-2',4'-diphenylspiro[9H-fluorene-9,5'-[1,3]thiazolidine] (**7c**, unstable primary cycloadduct).

¹H-NMR: 2.05 (s, MeN); 4.98, 6.06 (2s, H–C(4'), H–C(2')).

cis-3′-Methyl-2′,4′-diphenylspiro[9H-fluorene-9,5′-[1,3]thiazolidine] (8a). Yield: 140 mg (35%). Colorless needles. M.p. 154 – 156° (MeOH/CH₂Cl₂). IR (KBr): 1493m, 1448s, 1244m, 1144m, 1024m, 959m, 745vs, 732vs, 701s. ¹H-NMR: 2.20 (s, MeN); 4.25, 5.18 (2s, H – C(4′), H – C(2′)); 6.74 – 8.21 (m, 18 arom. H). ¹³C-NMR: 39.2 (q, MeN); 66.2 (s, C(5′)); 75.3, 84.2 (2d, C(4′), C(2′)); 118.9, 119.6, 124.9, 126.8, 127.2, 127.5, 127.6, 127.9, 128.2, 128.3, 128.5, 128.8 (12d, 18 arom. CH); 134.9, 139.1, 139.6, 140.8, 146.9, 148.5 (6s, 6 arom. C). CI-MS (NH₃): 422 (15, [M + NH₃]+), 406 (97, [M + 1]+), 284 (41), 152 (100). Anal. calc. for C₂₈H₂₃NS (405.57): C 82.92, H 5.72, N 3.45, S 7.91; found: C 82.95, H 5.87, N 3.36, S 8.08.

Reaction with 2c. The reaction was complete after 13 h. After evaporation of the solvent, the thick oily residue was analyzed by 1 H-NMR (presence of two sets of s at 5.17 and 6.30, and at 4.73 and 5.77 ppm, attributed to H-C(4) and H-C(2) of 7d and 8b, resp.; ratio of isomers ca.7:3). After chromatographic separation (SiO₂; petroleum ether with increasing amounts of CH₂Cl₂), a crystalline fraction identified as a 2:8 mixture of 7d and 8b was isolated. Attempted separation by fractional crystallization failed. In a second experiment, the crude

mixture of 7d and 8b was put on a SiO_2 column, and the separation was carried out only after 12 h. In this case, only isomerized cycloadduct 8b was isolated and purified by crystallization.

cis-3'-Isopropyl-2',4'-diphenylspiro[9H-fluorene-9,5'-[1,3]thiazolidine] (8b). Yield: 182 mg (42%). Colorless crystals. M.p. $162-164^{\circ}$ (MeOH/CH₂Cl₂). IR (KBr): 1448s, 1384w, 1174m, 1155m, 748s, 731vs, 702vs. 1 H-NMR: 0.77, 0.98 (2d, J=7.1, 6.7, Me_2 CH); 3.19 (m, Me_2 CH); 4.73, 5.77 (2s, H-C(4'), H-C(2')); 6.67-8.19 (m, 18 arom. H). 13 C-NMR: 14.8, 21.8 (2q, Me_2 CH); 49.9 (d, Me_2 CH); 65.5 (s, C(5')); 68.2, 81.2 (2d, C(4'), C(2')); 118.8, 119.4, 124.7, 126.5, 126.6, 127.0, 127.3, 127.4, 127.9, 128.0, 128.2, 128.3, 128.6 (13d, 18 arom. CH); 136.1, 139.2, 140.7, 141.9, 147.7, 148.8 (6s, 6 arom. C). CI-MS (NH₃): 434 (100, $[M+1]^+$), 400 (17), 312 (23), 180 (7), 148 (15), 106 (7). Anal. calc. for C_{30} H₂₇NS (433.61): C 83.10, H 6.28, N 3.23, S 7.39; found: C 83.14, H 6.27, N 3.23, S 7.53.

Reaction with 2d. The reaction was complete after 1 h, and the crude mixture was purified by chromatography according to the procedure described for 2b.

cis-3'-Benzyl-2',4'-diphenylspiro[9H-fluorene-9,5'-[1,3]thiazolidine] (8d). Yield: 212 mg (44%). Colorless needles. M.p. $161-162^\circ$ (MeOH). IR (KBr): 1493m, 1448s, 1119m (br.), 1067m, 1029m, 748s, 700s. 1 H-NMR: 3.89 (s, CH₂); 4.40, 5.45 (2s, H—C(4'), H—C(2')); 6.58-8.20 (m, 23 arom. H). 13 C-NMR: 52.0 (t, CH₂); 65.9 (s, C(5')); 69.6, 79.7 (2d, C(4')), C(2')); 119.2, 119.8, 125.1, 127.0, 127.2, 127.5, 127.7, 127.9, 128.1, 128.2, 128.4, 128.7, 128.8, 129.2, 129.4, 130.9 (16d, 23 arom. CH); 133.7, 134.9, 139.5, 139.8, 141.1, 147.6, 149.1 (7s, 7 arom. C). CI-MS (NH₃): 483 (43), 482 (100, $[M+1]^+$), 478 (16), 392 (7), 360 (5), 212 (6), 196 (10). Anal. calc. for $C_{34}H_{27}$ NS (481.66): C 84.78, H 5.65, N 2.91, S 6.66; found: C 84.89, H 5.64, N 2.90, S 6.54.

3.4. Reactions of 2c and 2e with Dimethyl Azodicarboxylate (9). A soln. of 9 (175 mg, 1.2 mmol) and 1 mmol of the corresponding 2 in abs. toluene (2 ml) was heated under reflux.

Reaction with 2c. After 11 h, additional 9 (29 mg, 0.2 mmol) was added to the mixture, and heating was continued for 4 h (15 h totally). After evaporation of the solvent, the oily residue was worked up by chromatography (Al₂O₃; petroleum ether with increasing amounts of CH₂Cl₂). The main fraction was obtained with 50% of CH₂Cl₂, and 10a was isolated as a thick, colorless oil. Attempts to obtain a crystalline product were not successful.

Dimethyl trans-4-Isopropyl-3,5-diphenyl-1,2,4-triazolidine-1,2-dicarboxylate (10a). Yield: 276 mg (72%). Colorless, thick oil. IR (neat): 1716s (br., C=O), 1444s, 1335vs (br.), 1209s (br.), 1124s, 1043m, 1028m, 972w, 827w, 760vs, 710s. 1 H-NMR: 1.06, 1.29 (2d, J = 6.3, 6.5, M = 2CH); 3.17 (m, Me₂CH); 3.96 (s, 2 MeO); 6.33 (s, H-C(3), H-C(5)); 7.53-7.79 (m, 10 arom. H). 1 3C-NMR: 21.2, 22.9 (2q, M = 2CH); 47.1 (d, Me₂CH); 53.3, 53.4 (2q, 2 MeO); 78.3, 78.4 (2d, C(3), C(5)); 127.3, 128.3, 128.4 (3d, 10 arom. CH); 138.0 (s, 2 arom. C); 157.4 (s, 2 C=O). Anal. calc. for C₂₁H₂₅N₃O₄ (383.45): C 65.78, H 6.57, N 10.96; found: C 65.25, H 6.29, N 10.88.

Reaction with 2e. After heating for 2 h, 2d was completely consumed. The solvent was evaporated, and the solid residue was analyzed by ${}^{1}H$ -NMR, which showed the presence of 10b as the sole product (s at 6.78 ppm). The crude product was triturated with a small amount of EtOH, and, after filtration, the crystalline product was purified by recrystallization.

Dimethyl trans-4-(Methoxyphenyl)-3,5-diphenyl-1,2,4-triazolidine-1,2-dicarboxylate (10b). Yield: 320 mg (71%). Colorless crystals. M.p. $156-161^\circ$ (EtOH/CH₂Cl₂) ([12]: $149-151^\circ$). IR (KBr): 1716vs (C=O), 1515s, 1444s, 1352vs, 1295s, 1276s, 1252vs, 1125s, 1038s, 819m, 763m, 706s. 1 H-NMR: 3.48 (br. s, 2 MeO); 3.63 (s, MeO); 6.35, 6.62 (AA'BB', J=8.5, 4 arom. H); 6.85 (s, H–C(3), H–C(5)); 7.28-7.37 (m, 10 arom. H). 13 C-NMR: 53.5, 55.4 (2q, 2 MeO); 75.2 (d, C(3), C(5)); 114.7, 114.9, 126.7, 128.58, 128.62 (5d, 14 arom. CH); 135.7, 137.5, 151.9 (3s, 4 arom. C); 157.5 (s, 2 C=O). CI-MS (NH₃): 449 (7), 448 (25, $[M+1]^+$), 254 (10), 238 (13), 237 (100), 212 (23). Anal. calc. for $C_{28}H_{25}N_3O_5$ (447.49): C 67.10, H 5.63, N 9.39; found: C 67.03, H 5.73, N 9.35.

3.5. Reactions of **2c** and **2e** with Dimethyl Acetylenedicarboxylate (**11**). A soln. of **11** (171 mg, 1.2 mmol) and 1.0 mmol of the corresponding **2** in 2 ml of abs. toluene was heated under reflux.

Reaction with **2c**. After 8-h heating, the solvent was evaporated, and the crude residue was analyzed by 1 H-NMR, which revealed the presence of ca. 50% of unconverted **2c**. The cycloadduct **12a** and 1H-pyrrole **13a** were present in a ratio of ca. 85:15 (based on the intensities of the Me_2 CH d between 0.70 and 1.15 ppm). Chromatographic separation (prep. TLC; SiO₂; hexane/CH₂Cl₂1:1) led to **12a** as the less polar fraction and **13a** as the more polar one. The products were purified by crystallization from MeOH.

In a similar experiment, the mixture was heated for 24 h. After typical workup, the ¹H-NMR spectrum of the crude residue revealed the presence of **13a** as the sole product. The product was triturated with a small amount of MeOH, filtered, and purified by recrystallization from MeOH.

Dimethyl trans-2,5-Dihydro-1-isopropyl-2,5-diphenyl-1H-pyrrole-3,4-dicarboxylate (**12a**). Yield (after 8 h): 133 mg (35%). Colorless crystals. M.p. 100–102° (MeOH). IR (KBr): 1744vs (C=O), 1721vs (C=O), 1455m, 1433m, 1317s, 1287m, 1244s, 1198s, 1031s, 750m, 706s. ¹H-NMR: 0.69, 0.82 (2d, Me₂CH); 2.87 (m, Me₂CH); 3.56

(s, 2 MeO); 5.61 (s, H-C(2), H-C(5)); 7.23-7.42 (m, 10 arom. H). 13 C-NMR: 21.5, 21.7 (2q, Me_2 CH); 47.1 (d, Me $_2$ CH); 51.8 (q, 2 MeO); 71.7 (d, C(2), C(5)); 127.7, 128.1, 128.2 (3d, 10 arom. CH); 139.6 (s, 2 arom. C); 140.8 (s, C=C); 163.7 (s, 2 C=O). ESI-MS (MeOH): 380 (100, [M + 1] $^+$). Anal. calc. for C $_{23}$ H $_{25}$ NO $_{4}$ (379.45): C 72.80, H 6.65, N 3.69; found: C 72.61, H 6.79, N 3.57.

Dimethyl 1-Isopropyl-2,5-diphenyl-1H-pyrrole-3,4-dicarboxylate (13a). Yield (after 24 h): 318 mg (84%). Colorless crystals. M.p. $174-176^{\circ}$ (MeOH). IR (KBr): 1716vs (C=O), 1482s, 1443s, 1341m, 1253s, 1199vs, 1169vs, 1061m, 766m, 703m. 1 H-NMR: 1.11 (d, J=7.1, Me_2 CH); 3.59 (s, 2 MeO); 4.24 (m, Me_2 CH); 7.42 (br. s, 10 arom. CH). 13 C-NMR: 23.2 (q, Me_2 CH); 50.0 (d, Me_2 CH); 51.3 (q, 2 MeO); 127.8, 128.6, 131.2 (3d, 10 arom. CH); 114.7, 131.8, 136.2 (3s, 6 arom. C); 165.3 (s, 2 C=O). EI-MS: 377 (100, M^{+*}), 346 (17), 335 (67), 304 (72), 303 (49), 272 (19).

Reaction with **2e**. The reaction was complete after 1 h. After evaporation of the solvent, the ¹H-NMR spectrum of the crude residue established the ratio **12b/13b** as *ca*. 2:1 (based on the intensities of the of MeO *s* at 3.62 and 3.71 ppm for **12b** and **13b**, resp.). The solvent was evaporated, and the semi-solid residue was dissolved in hot acetone. After cooling to r.t., the material crystallized was filtered and identified as **12b**. After one night in the refrigerator, the mother liquor yielded colorless crystals of **13b**. Anal. pure samples were obtained after recrystallization.

Dimethyl trans-2,5-Dihydro-1-(4-methoxyphenyl)-2,5-diphenyl-1H-pyrrole-3,4-dicarboxylate (12b). Yield: 186 mg (42%). Yellow crystals. M.p. $216-219^{\circ}$ (acetone). IR (KBr): 1736s (C=O), 1720s (C=O), 1512s, 1454m, 1436m, 1343m (br.), 1253s (br.), 1102m, 1030m, 815s, 759m, 703m. 1 H-NMR: 3.56 (s, MeO); 3.62 (s, 2 MeO); 6.22 (s, H-C(2), H-C(5)); 6.37, 6.51 (AA'BB', J=8.8, 4 arom. CH); 720-7.34 (m, 10 arom. CH). 13 C-NMR: 52.0 (g, 2 MeO); 55.2 (g, MeO); 71.1 (g, C(2), C(5)); 114.1, 115.3, 127.3, 128.0, 128.6 (128.6) (128.6), 128.6 (128.6), 128.6), 128.6 (128.6), 128.6), 128.6, 128.6), 128.6, 128.6, 128.6, 128.6), 128.6

Dimethyl 1-(4-Methoxyphenyl)-2,5-diphenyl-IH-pyrrole-3,4-dicarboxylate (13b). Yield: 102 mg (23%). Colorless crystals. M.p. $196-198^\circ$ (acetone; [28]: $210-211^\circ$). IR (KBr): 1731vs (C=O), 1712vs (C=O), 1515s, 1483s, 1445s, 1254s, 1212s, 1182s, 1102m, 1061m, 763m, 701m. $^1\text{H-NMR}$: 3.66 (s, MeO); 3.71 (s, 2 MeO); 6.58, 6.74 (AA'BB', J=8.9, 4 arom. CH); 7.17-7.25 (m, 10 arom. CH). $^{13}\text{C-NMR}$: 51.6 (q, 2 MeO); 55.1 (q, MeO); 113.6, 127.5, 127.9, 129.7, 130.7 (5d, 14 arom. CH); 114.6, 130.4, 136.8, 158.6 (4s, 8 arom. C); 165.5 (s, 2 C=O). CI-MS: 442 (100, $[M+1]^+$), 427 (7), 306 (13). Anal. calc. for $C_{27}H_{23}NO_5$ (441.48): C 73.46, H 5.25, N 3.17; found: C 73.53, H 5.25, N 3.14.

3.6. Reactions of 2c and 2e with Dimethyl Funarate (14). A soln. of 14 (173 mg, 1.2 mmol) and 1.0 mmol of the corresponding 2 in 2 ml of abs. toluene was heated under reflux. The progress of the reaction was followed by TLC.

Reaction with 2c. After heating for 18 h, the solvent was evaporated, and the crude residue was analyzed by ¹H-NMR. Based on the intensities of the Me₂CH m at 2.70 and 3.02 ppm, the ratio 15a/16a was established as ca. 1:1. After evaporation of CDCl₃, the residue was dissolved in hot MeOH, cooled to r.t., and kept overnight in the refrigerator. Colorless crystals of 16a were filtered, and the mother liquor was evaporated. The semi-solid residue was separated by prep. TLC (SiO₂; hexane/CH₂Cl₂). The less polar main fraction was isolated and identified as 15a, and some additional 16a was obtained as the more polar fraction.

Dimethyl trans,trans,trans-1-Isopropyl-2,5-diphenylpyrrolidine-3,4-dicarboxylate (**15a**). Yield: 88 mg (23%). Colorless crystals. M.p. $42-45^\circ$ (hexane). IR (KBr): 1732vs (C=O), 1456m, 1436m, 1263m, 1206vs, 174s, 1005w, 762m, 704s. ¹H-NMR: 0.27, 0.83 (2d, J=6.7, 6.9, Me_2 CH); 2.70 (m, Me_2 CH); 3.42 (m, 2 H); 3.57 (s, 2 MeO); 4.72 (m, 2 H); 7.13 – 7.34 (m, 10 arom. CH). ¹³C-NMR: 18.8, 20.6 (2q, Me_2 CH); 46.6 (d, Me_2 CH); 52.0, 52.1 (2q, 2 MeO); 55.0, 67.1 (2d, C(2), C(3), C(4), C(5)); 127.4, 128.2, 128.3 (3d, 5 arom. CH); 143.8 (s, 2 arom. C); 173.6 (s, 2 C=O). CI-MS (NH₃): 382 (100, [M+1]+). Anal. calc. for C_{23} H₂₇NO₄ (381.47): C 72.42, H 7.13, N 3.67; found: C 72.04, H 6.91, N 3.50.

Dimethyl cis,trans,cis-1-Isopropyl-2,5-diphenylpyrrolidine-3,4-dicarboxylate (16a). Yield: 88 mg (23%). Colorless crystals. M.p. 111 – 113° (MeOH). IR (KBr): 1744vs (C=O), 1732vs (C=O), 1454m, 1440m, 1209vs, 1172s, 1006m, 703s. 1 H-NMR: 0.80, 0.82 (2d, J = 6.7, 7.0, Me_2 CH); 3.02 (m, Me₂CH); 3.26 (s, 2 MeO); 4.10 (m, 2 H); 5.01 (m, 2 H); 7.20 – 7.30 (m, 10 arom H). 13 C-NMR: 21.7, 22.8 (2q, Me_2 CH); 48.3 (d, Me₂CH); 49.4 (d, C(3), C(4)); 51.2 (q, 2 MeO); 65.8 (d, C(2), C(5)); 127.3, 127.8, 127.9 (3d, 10 arom. CH); 141.8 (s, 2 arom. C); 170.7 (s, 2 C=O). EI-MS: 381 (<1, M^{++}), 366 (100, [M – Me] $^{+}$), 338 (44), 194 (41), 150 (24). Anal. calc. for C₂₃H₂₇NO₄ (381.47): C 72.42, H 7.13, N 3.67; found: C 72.34, H 7.20, N 3.54.

Reaction with 2e. The reaction was complete after 12 h. The solvent was evaporated, and the crude residue was analyzed by 1 H-NMR. Based on the intensities of signals at 5.66 (m, H-C(2), H-C(5)) in 15b) and

5.44 ppm (m, H-C(2), H-C(5) in 16b), the ratio 15b/16b was established as ca. 1:2. Attempted separation of the diastereoisomers by fractional crystallization failed. Chromatography (prep. TLC; SiO₂; hexane/CH₂Cl₂ 1:4) led to 15b as the less polar fraction and 16b as the more polar one. Recrystallization gave anal. pure samples.

Dimethyl trans,trans,trans-1-(4-Methoxyphenyl)-2,5-diphenylpyrrolidine-3,4-dicarboxylate (15b). Yield: 334 mg (75%). M.p. $132-134^{\circ}$ (MeOH/CH₂Cl₂). IR (KBr): 1732s (C=O), 1513s, 1455m, 1438m, 1244vs, 1038m, 816m, 763m, 703m. 1 H-NMR: 3.54 (s, 2 MeO); 3.54-3.59 (m, H-C(3), H-C(4)); 3.59 (s, MeO); 5.65-5.67 (m, H-C(2), H-C(5)); 6.32, 6.52 (AA'BB', J=9.1, 4 arom. H); 7.15-7.27 (m, 10 arom. H). 13 C-NMR: 52.2 (q, 2 MeO); 55.2 (q, MeO); 55.3 (d, C(3), C(4)); 66.0 (d, C(2), C(5)); 113.9, 117.0, 126.6, 127.1, 128.4 (5d, 14 arom. CH); 138.2, 141.7, 151.3 (3s, 4 arom. C); 172.0 (s, 2 C=O). CI-MS (NH₃): 447 (30), 446 (100, $[M+1]^{+}$), 236 (24), 222 (47). Anal. calc. for $C_{27}H_{27}NO_5$ (445.51): C 72.79, H 6.11, N 3.14; found: C 72.73, H 6.05, N 3.06

Dimethyl cis,trans,cis-1-(4-Methoxyphenyl)-2,5-diphenylpyrrolidine-3,4-dicarboxylate (**16b**). Yield: 85 mg (19%). Colorless crystals. M.p. $263-265^{\circ}$ (MeOH/CH₂Cl₂). IR (KBr): 1733vs (C=O), 1511vs, 1454m, 1437m, 1297m, 1259s, 1239s, 1219s, 1177vs, 1041m, 814m, 755m, 704m. 1 H-NMR: 3.34 (s, 2 MeO); 3.52 (s, MeO); 4.07 (m, H-C(3), H-C(4)); 5.44 (m, H-C(2), H-C(5)); 6.20, 6.48 (AA'BB', J=8.5, 4 arom. H); 7.11-7.25 (m, 10 arom. H). 13 C-NMR: 48.7 (d, C(3), C(4)); 51.6 (q, 2 MeO); 55.3 (q, MeO); 63.9 (d, C(2), C(5)); 114.4, 126.9, 127.8, 128.4 (4d, 14 arom. CH); 137.8, 138.9, 151.0 (3s, 4 arom. C); 169.7 (s, 2 C=O). Anal. calc. for $C_{27}H_{27}$ NO₅ (445.51); C 72.79, H 6.11, N 3.14; found: C 72.73, H 6.05, N 3.06.

3.7. Reaction of 2d with Dimethyl Maleate (17). A soln. of 17 (173 mg, 1.2 mmol) and 2d (285 mg, 1.0 mmol) in abs. toluene (2 ml) was heated under reflux. After 24 h, no 2d was detected (TLC). The solvent was evaporated, and the crude residue was analyzed by ¹H-NMR showing the presence of 18/19 in a ratio of 2:1. After evaporation of CDCl₃, the solid residue was dissolved in hot EtOH. After cooling to r.t., the colorless precipitate was filtered and identified as 18. The mother liquor was kept in the refrigerator, and, the next day, crystalline 19 was filtered. Recrystallization of the crude products led to anal. pure compounds.

Dimethyl cis,cis,trans-1-(4-Methoxyphenyl)-2,5-diphenylpyrrolidine-3,4-dicarboxylate (**18**). Yield: 192 mg (43%). Colorless crystals. M.p. $197-199^{\circ}$ (EtOH/CH₂Cl₂). IR (KBr): 1744s (C=O), 1514vs, 1454m, 1438m, 1243s (br.), 1217s (br.), 1178m, 1039m, 815w, 700m. ¹H-NMR: 3.18, 3.54, 3.62 (3s, 3 MeO); 3.40, 3.87 (2dd, H-C(3), H-C(4)); 5.60, 5.84 (AB, J=7.2, H-C(2), H-C(5)); 6.27, 6.41 (AA'BB', J=9.0, 4 arom. H); 7.10-7.41 (m, 10 arom. H). ¹³C-NMR: 51.3, 51.9, 52.8 (3q, 3 MeO); 55.1 (d, C(3), C(4)); 65.8, 66.7 (2d, C(2), C(5)); 113.3, 118.7, 126.9, 127.4, 127.5, 127.8, 127.9, 128.2 (8d, 14 arom. CH); 137.9, 138.2, 143.3, 151.2 (4s, 4 arom. C); 171.0 (s, 2 C=O). CI-MS (NH₃): 447 (28), 446 (100, [M+1]+), 444 (5). Anal. calc. for C₂₇H₂₇NO₅ (445.51): C 72.79, H 6.11, N 3.14; found: C 72.61, H 6.15, N 3.12.

Dimethyl trans-4,5-Dihydro-1-(4-methoxyphenyl)-2,5-diphenyl-1H-pyrrole-3,4-dicarboxylate (19). Yield: 75 mg (17%). Colorless crystals. M.p. $152-154^{\circ}$ (EtOH/CH₂Cl₂). IR (KBr): 2949m, 1736vs, 1674s, 1611s, 1593s, 1573s, 1509s, 1437s, 1363s (br.), 1210vs (br.), 1114s, 1055s, 1030s, 840s, 765s, 696s. 1 H-NMR: 3.53, 3.62, 3.79 (3s, 3 MeO); 3.94, 4.96 (AB, J = 5.5, H – C(4), H – C(5)); 6.51, 6.62 (AA'BB', J = 9.0, 4 arom. H); 7.22 – 7.48 (m, 10 arom. H). 13 C-NMR: 50.5, 52.3, 55.1 (3q, 3 MeO); 56.5, 73.0 (2d, C(4), C(5)); 99.1 (s, C(3)); 113.8, 126.3, 126.7, 127.5, 128.0, 128.9, 129.2, 130.0 (8d, 14 arom. CH); 161.0 (s, C(2)); 131.0, 135.5, 142.0, 156.9 (4s, 4 arom. C); 165.5, 174.2 (2s, 2 C=O). CI-MS (NH₃): 444 (100, [M + 1] $^+$). Anal. calc. for C₂₇H₂₅NO₅ (443.50): C 73.12, H 5.68, N 3.16; found: C 73.05, H 5.69, N 3.15.

4. Crystal-Structure Determination of **7b**, **8a**, **15b**, and **16b**¹²). All measurements for **7b** were made on a KM-4 Kuma diffractometer [29] with graphite monochromatized CuK_a radiation (λ 1.54178 Å). For the other compounds, the measurements were conducted on a Nonius KappaCCD area-detector diffractometer [30] with graphite-monochromated MoK_a radiation (λ 0.71073 Å) and an Oxford Cryosystems Cryostream 700 cooler. The data collection and refinement parameters are given in the Table, views of the molecules are shown in Figs. I-3. Data reduction for **7b** was performed with Dataproc [31], while HKL Denzo and Scalepack [32] was used for **8a**, **15b**, and **16b**. The intensities were corrected for Lorentz and polarization effects. Anal. [33] and empirical (multi-scan method [34]) absorption corrections were applied for **7b** and **8a**, resp. Each structure was solved by direct methods [35][36], which revealed the positions of all non-H-atoms. The non-H-atoms were

¹²⁾ CCDC-198065 and 208506-208508 contain the supplementary crystallographic data for this paper. These data can be obtained, free of charge, via www.ccdc.cam.ac.uk/conts/retrieving.html (or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, U.K.; fax: +441223336033; e-mail: deposit@ccdc.cam.ac.uk).

Table. Crystallographic Data for Compounds 7b, 8a, 15b, and 16b

	7b	8a	15b	16b
Crystallized from	EtOH	МеОН	MeOH/CH ₂ Cl ₂	MeOH/CH ₂ Cl ₂
Empirical formula	$C_{30}H_{29}NS$	$C_{28}H_{23}NS$	$C_{27}H_{27}NO_5$	$C_{27}H_{27}NO_5$
Formula weight	435.60	405.56	445.51	445.51
Crystal color, habit	colorless, prism	colorless, plate	colorless, prism	colorless, tablet
Crystal dimensions [mm]	$0.20\times0.30\times0.50$	$0.07 \times 0.30 \times 0.30$	$0.15\times0.20\times0.27$	$0.12\times0.15\times0.18$
Temp. [K]	293(1)	160(1)	160(1)	160(1)
Crystal system	triclinic	orthorhombic	monoclinic	monoclinic
Space group	$P\bar{1}$	Pbca	Cc	$P2_{1}/c$
Z	4	16	4	4
Reflections for cell determination	54	64406	3469	5290
2θ Range for cell determination [°]	10 - 22	4-55	4 - 60	4-55
Unit cell parameters a [Å]	11.768(2)	12.9886(1)	12.4880(2)	11.0645(2)
<i>b</i> [Å]	15.276(3)	22.8639(2)	14.7967(3)	20.1040(4)
c [Å]	15.660(3)	28.9652(3)	13.8734(3)	10.7831(2)
α $[^{\circ}]$	73.90(2)	90	90	90
eta [$^{\circ}$]	71.04(2)	90	116.6317(8)	110.4783(8)
γ [°]	71.87(2)	90	90	90
$V [\mathring{\mathbf{A}}^3]$	2481.1(8)	8601.8(1)	2291.57(8)	2247.02(7)
D_x [g cm ⁻³]	1.166	1.253	1.291	1.317
Linear absorption coefficient [mm ⁻¹]	1.27	0.165	0.0888	0.0906
Transmission factors [min; max]	0.592; 0.802	0.893; 0.992	_	_
Scan type	ω	ϕ and ω	ϕ and ω	ω
$2\theta_{(\max)}$ [°]	135	55	60	55
Total reflections measured	8624	89971	27716	54079
Symmetry-independent reflections	8293	9845	3359	5131
Reflections with $I > 2\sigma(I)$	5705	6209	2828	3831
Reflections used in refinement	8293	9844	3358	5129
Parameters refined; restraints	589	544	302; 2	302
Final $R(F)$ $(I > 2\sigma(I)$	0.0546	0.0468	0.0415	0.0438
reflections)				
$wR(F^2)$ (all data)	0.1887	0.1249	0.0995	0.1154
Weighting parameters $[a; b]^a$)	0.1264; 0.1241	0.0625; 0.4740	0.0053; 0.1625	0.0574; 0.4651
Goodness-of-fit	1.112	1.016	1.036	1.045
Secondary extinction coefficient		0.0011(2)	0.012(2)	0.012(2)
Final Δ_{\max}/σ	0.003	0.001	0.001	0.001
$\Delta \rho \text{ (max; min) [e Å}^{-3}]$	0.88; -0.35	0.27; -0.43	0.24; -0.19	0.20; -0.26

a) $w^{-1} = \sigma^2(F_0^2) + (aP)^2 + bP$, where $P = (F_0^2 + 2F_0^2)/3$

refined anisotropically. All of the H-atoms were placed in geometrically calculated positions and refined with a riding model where each H-atom was assigned a fixed isotropic displacement parameter with a value equal to $1.2U_{\rm eq}$ of its parent C-atom $(1.5U_{\rm eq}$ for Me groups). In both **7b** and **8a**, there are two symmetry-independent molecules in the asymmetric unit. The atomic coordinates of the two molecules were tested carefully for a relationship from a higher symmetry space group with the program PLATON [37], but none could be found. One Me C-atom in molecule B of **7b**, C(33B), is disordered. Two positions were defined for this atom and the site occupation factor of the major component refined to 0.535(14). Refinement of each structure was carried out on F^2 by full-matrix least-squares procedures, which minimized the function $\sum w \left(F_{\rm o}^2 - F_{\rm c}^2\right)^2$. For **8a**, **15b**, and **16b**, corrections for secondary extinction were applied, and one, one, and two reflections, resp., whose intensities were considered to be extreme outliers, were omitted from the final refinements.

Neutral-atom-scattering factors for the non-H-atoms were taken from [38a], and the scattering factors for H-atoms were taken from [39]. Anomalous dispersion effects were included in F_c [40]; the values for f' and f''

were those of [38b]. The values of the mass-attenuation coefficients are those of [38c]. All calculations were performed using SHELXL97 [41].

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